



AEC-NASA TECH BRIEF



AEC-NASA Tech Briefs describe innovations resulting from the research and development program of the U.S. AEC or from AEC-NASA interagency efforts. They are issued to encourage commercial application. Tech Briefs are published by NASA and may be purchased, at 15 cents each, from the Clearinghouse for Federal Scientific and Technical Information, Springfield, Virginia 22151.

Direct Measurement of $^{14}\text{CO}_2$ by Liquid Scintillation Counting

The problem:

To develop a technique for the direct measurement of carbon-14 in carbon dioxide. Previous methods suffered from such drawbacks as nonquantitative absorption or reaction of the CO_2 , quenching from the addition of absorbents or reactants, and isotope effects during successive reactions.

The solution:

A liquid scintillation counting technique for measurement of ^{14}C activity. The carbonaceous sample is first converted into CO_2 , and the $^{14}\text{CO}_2$ is then measured by a liquid scintillation counter. This method has high counting efficiency and eliminates many of the basic problems encountered with previous measuring techniques.

How it's done:

Three different counting systems have been used with equal success: a room temperature single-multiplier phototube (RCA-8575) system; a -20°C single-multiplier phototube (DuMont-6292) system; and a 0°C coincidence liquid scintillation counter.

Two scintillator solutions were used to match emission from the solution with response of the multiplier phototubes. With the RCA-8575 multiplier phototube, a solution of 3 g of PPO in a liter of toluene was used; with the DuMont-6292 multiplier phototube, a solution of 7 g of PPO and 0.5 g of $\text{M}_2\text{-POPOP}$ in a liter of toluene was used.

The CO_2 gas samples were prepared by normal vacuum line manipulations. The CO_2 was condensed with liquid nitrogen onto a previously degassed and frozen volume of the scintillator solution. The sample tubes were then sealed by flame, minimizing the volume of space above the scintillator solution.

The solubility of CO_2 in toluene base scintillator solution is sufficient to allow the ^{14}C measurement at room temperature. By the use of a special type of sample container, the space above the solution was minimized to the extent that greater than 99% of the CO_2 was dissolved in the solution.

To determine the concentration quenching of CO_2 , samples were prepared with varying concentrations of CO_2 . The relative scintillation yields were determined by the Compton edge technique. There was no detectable quenching up to 5 cc (STP) per ml of liquid scintillator solution.

To determine the counting efficiency, several samples of known volume of ^{14}C containing CO_2 were dissolved in 2 ml of scintillator solution and counted in a room temperature single-multiplier phototube system. The counting rates were determined by the integral counting technique. Aliquots of the same gas were counted in a gas counter. The extrapolated integral counting rates of samples in the liquid scintillators agreed with $\pm 2\%$ of the absolute disintegration rate obtained with the gas counter.

This method can be useful for either assay or absolute measurement of ^{14}C radioactivities in samples converted into CO_2 . Counting efficiencies of greater than 90% are obtained with low backgrounds. With an integral counting technique, a counting efficiency of $(100 \pm 2)\%$ was obtained. The lower level of detection of ^{14}C by this method is 10^{-7} Ci per cc (STP) of CO_2 for a 10 cc sample (about 2 disintegrations per minute).

Notes:

1. The use of scintillation counters is not new, but this particular measurement is a novel application.

(continued overleaf)

2. The technique can be used to achieve a percent substitution reaction, and may be of interest to university and hospital laboratories as an analytical technique.
3. Inquiries concerning this report may be directed to:

Office of Industrial Cooperation
Argonne National Laboratory
9700 South Cass Avenue
Argonne, Illinois 60439
Reference: B69-10092

Source: D. L. Horrocks
Chemistry Division
(ARG-10237)

Patent status:

Inquiries about obtaining rights for commercial use of this innovation may be made to:

Mr. George H. Lee, Chief
Chicago Patent Group
U. S. Atomic Energy Commission
Chicago Operations Office
9800 South Cass Avenue
Argonne, Illinois 60439